## Fluid Mechanics I

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### References



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# Fluid Mechanics Instances Life adventure Environment behavior Veather Daily Life Animals behavior Industries And ...







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- Temperature T is a measure of the internal energy level of a fluid.
- · It may vary considerably during high-speed flow of a gas.
- Although engineers often use Celsius or Fahrenheit scales for convenience, many applications require *absolute* (Kelvin or Rankine) temperature scales: (°R = °F + 459.69 K = °C 273.16)
- If temperature differences are strong, *heat transfer may be important.*

### Density

- · The density of a fluid, is its mass per unit volume.
- Density is highly variable in gases and increases nearly proportionally to the pressure level.
- Density in liquids is nearly constant; (water about 1000 kg/m<sup>3</sup>) increases only 1% if the pressure is increased by a factor of 220. (most liquid flows are treated analytically as "incompressible.")
- In general, liquids are about 3 orders of magnitude more dense than gases at atmospheric pressure.
- The heaviest liquid: mercury, the lightest gas: hydrogen They differ by a factor of 162,000!
- Various physical Properties => dimensional analysis

### Specific Weight

- The specific weight of a fluid, is its weight per unit volume.
- · The units of are weight per unit volume, in lbf/ft^3 or N/m^3.
- In standard earth gravity, the specific weights of air and water at 20°C and 1 atm are approximately:
  - air (1.205 kg/m^3)\*(9.807 m/s^2) 11.8 N/m^3 0.0752 lbf/ft^3 water (998 kg/m^3)\*(9.807 m/s^2) 9790 N/m^3 62.4 lbf/ft^3
- Specific weight is very useful in the hydrostatic-pressure applications.

### Specific Gravity

• Specific gravity, is the ratio of a fluid density to a standard reference fluid, water (for liquids), and air (for gases):

$$SG_{gas} = \frac{\rho_{gas}}{\rho_{air}} = \frac{\rho_{gas}}{1.205 \text{ kg/m}^3}$$
$$SG_{liquid} = \frac{\rho_{liquid}}{\rho_{water}} = \frac{\rho_{liquid}}{998 \text{ kg/m}^3}$$

• Engineers find these dimensionless ratios easier to remember than the actual numerical values of density of a variety of fluids.

## Potential and Kinetic Energies In thermostatics the only energy in a substance is that stored in a system by molecular activity and molecular bonding forces (*internal energy*) *fluid flow:* 2 more energy terms which arise from newtonian mechanics: the potential energy and kinetic energy. The potential energy equals the work required to move the system of mass *m from* the origin to a position against a gravity field g. The kinetic energy equals the work required to change the speed of

- the mass from zero to velocity V  $e = a + \frac{1}{2}V^2 + gz$ • The molecular internal energy is a function of T and p for the single-
  - The molecular internal energy is a function of T and p for the singlephase pure substance, whereas the potential and kinetic energies are kinematic properties.















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 $x_{\rm CP} = -\frac{I_{xy}\sin\theta}{h_{\rm CG}A}$ 

















































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The Energy Equation
$\beta = dE/dm = e.  \frac{dQ}{dt} - \frac{dW}{dt} = \frac{dE}{dt} = \frac{d}{dt} \left( \int_{CV} e\rho \ dV \right) + \int_{CS} e\rho(\mathbf{V} \cdot \mathbf{n}) \ dA$
$e = e_{\text{internal}} + e_{\text{kinetic}} + e_{\text{potential}} + e_{\text{other}}$ $\dot{w} = \dot{w}_{\text{shaff}} + \dot{w}_{\text{press}} + \dot{W}_{\text{viscous stresses}} = \dot{W}_s + \dot{W}_p + \dot{W}_v$
$d\dot{W}_p = -(p \ dA)V_{n,\mathrm{in}} = -p(-\mathbf{V}\cdot\mathbf{n}) \ dA \qquad d\dot{W}_v = -\tau\cdot\mathbf{V} \ dA$
The total pressure work is the integral over the control surface $\dot{W}_{\nu} = -\int_{-\infty} \tau \cdot \mathbf{V}  dA$
$\dot{W}_p = \int_{CS} p(\mathbf{V} \cdot \mathbf{n})  dA$
$\dot{W} = \dot{W}_s + \int_{CS} p(\mathbf{V} \cdot \mathbf{n})  dA - \int_{CS} (\tau \cdot \mathbf{V})_{SS}  dA$
$\dot{Q} - \dot{W}_s - (\dot{W}_v)_{\rm SS} = \frac{\partial}{\partial t} \left( \int_{\rm CV} ep \ dV \right) + \int_{\rm CS} \left( e + \frac{p}{\rho} \right) \rho(\mathbf{V} \cdot \mathbf{n}) \ dA$
$\dot{Q} - \dot{W}_s - \dot{W}_v = \frac{\partial}{\partial t} \left[ \int_{CV} \left( \dot{u} + \frac{1}{2} V^2 + gz \right) \rho  dV \right] + \int_{CS} \left( \dot{h} + \frac{1}{2} V^2 + gz \right) \rho (\mathbf{V} \cdot \mathbf{n})  dA$
Friction Losses in Low-Speed Flow $\left(\frac{p}{\gamma} + \frac{V^2}{2g} + z\right)_{in} = \left(\frac{p}{\gamma} + \frac{V^2}{2g} + z\right)_{out} + h_{friction} - h_{pump} + h_{turbine}$